

PROTECTION AGAINST INFECTIONS

The control of microbial growth is necessary in many practical situations, and significant advances in agriculture, medicine, and food science have been made through study of this area of microbiology. The microorganisms are ubiquitous in nature. In order to study the nature and characteristics of a particular microbe, it is essential to isolate it from other contaminating microorganisms. This can be achieved by maintaining a completely sterile environment in which the microbe of interest is selectively grown. It is necessary that not only the place you are working with microorganisms should be free from contamination (other living organisms) but, the media and the materials you are using to handle and grow specific microorganisms should be free from other microbial contaminants. For this purpose 'sterilization' of the place of work materials and media have to be done.

"Control of growth" as used here means to prevent the growth of microorganisms. This control is affected in two basic ways: (1) by killing microorganisms or (2) by inhibiting the growth of microorganisms. Control of growth usually involves the use of physical or chemical agents which either kill or prevent the growth of microorganisms. Agents which kill cells are called **cidal** agents; agents which inhibit the growth of cells (without killing them) are referred to as **static** agents. Thus the term **bactericidal** refers to killing bacteria and bacteriostatic **refers** to inhibiting the growth of bacterial cells. A **bactericide** kills bacteria; a **fungicide** kills fungi, and so on.

Sterilization is a process of complete removal or killing of all forms of microbial life including spores from an object, surface, medium or environment without spoiling its nature.

Methods

There are various sterilization techniques available. However, several factors influence the effectiveness of sterilization process like, the concentration of antimicrobial agents, time and temperature of exposure, size of population, type of contaminating microbes etc.

Sterilization is brought about by a combination of physical and chemical agents that adversely affect the microorganisms either by causing damage to the cell wall or cell membrane or by inactivating the enzymes or by interfering with the synthesis of nucleic acids and protein.

I. PHYSICAL AGENTS

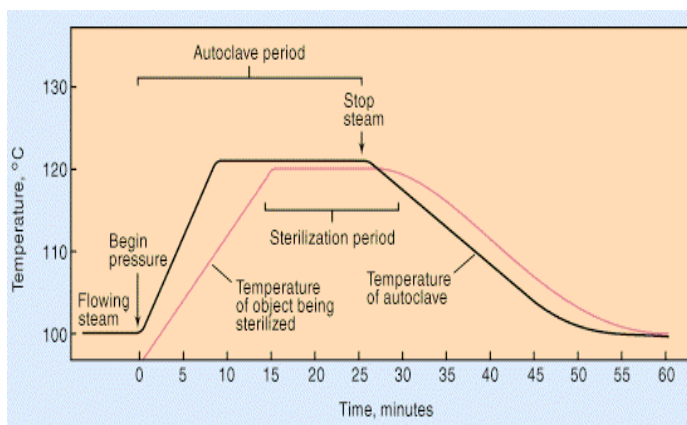
There are different types of physical agents.

(i) **Heat:** The heat employed for removal of micro-organisms varied with the nature of object and also depend on the purpose. Based on these different processes are employed.

(a) Moist heat

It is the widely used effective means of sterilization process. In this, steam under high pressure is employed which imparts high penetration power resulting in the hydration of cells and coagulation of protein leading to the death of the microorganism. Autoclave is the apparatus used for sterilization by moist heat.

The autoclave is a double-jacketed steam chamber. The chamber is equipped with a device for generating saturated steam. It can be maintained at a particular temperature and pressure for any period of time. During operation of autoclave the air in the chamber is evacuated by steam since presence of air will reduce the temperature in the chamber. The time required for sterilization will depend upon the materials to be sterilized. Solid materials must be heated for a longer



time (1-2 hours) while liquid media can be sterilized within 15-30 minutes. Also acidic materials require shorter period than alkali materials. A temperature of 121°C for 15 min at a pressure of 15 lbs/ sq.inch is the sterilizing condition in the autoclave.

Advantages

Steam can penetrate through materials and sterilization is achieved by the coagulation or denaturation of proteins and other cell constituents. Liquid media, solid media, laboratory equipments (cloth, glasswares, etc.,) can be sterilized. The temperature and pressure is high enough to kill spores, vegetative cells and viruses.

Disadvantages

Temperature sensitive media, animal tissue culture media, antibiotics, amino acids, cannot be sterilized. Sometimes water may get inside incase of improper packing.

(b) Dry heat

This process is accomplished in a hot-air oven. Hot air or dry heat is employed for sterilization. The dry heat penetrates substances more slowly than the moist heat. Hence, the time required for effective sterilization is long (2 to 3 hours) and also the temperature required is too high (160°C -180°C). Microbial death results from the oxidation of cell constituents.

Advantages

Dry heat does not corrode glassware and metal instruments as moist heat does. All glassware's can be sterilized.

Disadvantages

The sterilization process is slow. It is not suitable for heat sensitive materials like many plastic and rubber items.

(c) **Boiling** at 100°C for 30 minutes. Kills everything except some endospores (Actually, for the purposes of purifying drinking water 100⁰ for five minutes is probably adequate though there have been some reports that Giardia cysts can survive this process). To kill endospores, and therefore **sterilize** the solution, very long or **intermittent boiling** is required.

(d) **Pasteurization** is the use of mild heat to reduce the number of microorganisms in a product or food. In the case of pasteurization of milk the time and temperature depend on killing potential pathogens that are transmitted in milk, i.e., *staphylococci*, *streptococci*, *Brucella abortus* and *Mycobacterium tuberculosis*. For pasteurization of milk:

batch method (Low temperature holding): 62.8⁰C for 30 minutes flash

method (High temperature short time): 71.7⁰C for 15 seconds

(e) **Intermittent sterilization or Tyndallization** is the process of boiling the materials at 100⁰C for 30 min. successively for three consecutive days. Destroys vegetative cells and spores; germinated spores.

(f) **Incineration** burns organisms and physically destroys them. Incineration is the complete burning of the material in to ashes. Used for needles, inoculating wires, glassware, etc. and objects not destroyed in the incineration process. This is the direct and ultimate method of destroying cells. It is achieved by keeping the materials directly in contact with the flame of Bunsen burner as a result all the microorganisms in the surface are destroyed completely. Inoculating loops, needles and spreading rods are sterilized by this method.

Advantages: Immediate and quick.

Disadvantages: Cannot be used to sterilize heat labile material, material is lost by incineration.

Recommended use of heat to control bacterial growth

Treatment	Temperature	Effectiveness
Incineration	500 ⁰ C	Vaporizes organic material on non flammable Surfaces but may destroy many substances in the process
Boiling	100 ⁰ C	30 minutes of boiling kills microbial pathogens and vegetative forms of bacteria but may not kill bacterial
Intermittent boiling	100 ⁰ C	Three 30-minute intervals of boiling, followed by Periods of cooling kills bacterial endospores.
Autoclave and pressure cooker (steam under pressure)	121 ⁰ C for 15 min. at 15 lbs/ sq.inch pressure	Kills all forms of life including bacterial endospores. The substance being sterilized must be maintained at the effective T for the full time
Dry heat (hot air oven)	160 ⁰ C /2 hours	For materials that must remain dry and which are not destroyed at the between 121 ⁰ C and 170 ⁰ C Good for glassware, metal, not plastic or rubber items
Dry heat (hot air oven)	180 ⁰ C /1 hour	Same as above. Note increasing T by 10 degrees shortens the sterilizing time by 50 percent
Pasteurization	62.8 ⁰ C /30 min.	kills most vegetative bacterial cells including pathogens such as streptococci, staphylococci and
Pasteurization (flash method)	71.7 ⁰ C/15 seconds	Effect on bacterial cells similar to batch method; for milk, this method is more conducive to industry and has fewer undesirable effects on quality

(ii) Radiation

Energy transmitted through space in a variety of forms is generally called radiation. It is also known as "cold sterilization" as only little heat is produced during the process. The most significant of this is electromagnetic radiation. The energy content and radiation wavelength are inversely proportional to each

other. Radiation may be ionizing or non-ionizing.

Ionizing radiation

High-energy electron beams (Gamma, X-rays, alpha and beta particles) have sufficient energy to cause ionization of molecules. They drive away electrons and split the molecules into ions. Water molecules are split into hydroxyl radicals (OH⁻), electrons and hydrogen ions (H⁺). OH⁻ ions are highly reactive and destructive to normal cellular compounds such as DNA and proteins. Thus ionizing radiations are used in sterilization.

e.g. ³⁶Cs, ⁶⁰Co

Advantages: X-rays and Gamma rays have high penetrating power. Packed food and medical equipments are sterilized by using x-rays and gamma rays.

Disadvantage: Generating and controlling X-rays for sterilization is highly expensive.

Non-ionizing radiation

This includes ultraviolet (UV) rays. UV at a wavelength of 265 nm is most bactericidal. Absorption of UV radiation produces chemical modification of nucleoproteins i.e., thymine dimer formation that leads to misleading of genetic codes. This mutation impairs the total functions of the organism, consequently causing its death.

Advantages

It is used to maintain aseptic conditions in laminar air flow chamber, lab, hospitals, pharmaceuticals, industries etc., and also in the sterilization of water and air.

Disadvantage:

UV radiation has very little ability to penetrate matter and hence the micro organisms on the surface of an object are destroyed.

III) Filtration

Filtration involves the passage of liquid or gas through a screen like material that has spores small enough to retain the micro organism of certain size. It is used to sterilize heat sensitive substance like enzyme solutions, bacterial toxins, certain biological media, cell extract and some sugars. Various types of filters are available in different grades of porosity. Vacuum or pressure is required to move the solutes through the filter.

Involves the physical removal of all cells in a liquid or gas, especially important to sterilize solutions which would be denatured by heat (eg: antibiotics, injectable drugs, amino acids, vitamins etc.)

Advantages:

It is the best way to reduce microbial population in solutions of heat sensitive materials and it is used to sterilize liquid media, vitamin solutions, hormones, growth factors, enzymes.

Disadvantages

Pleomorphic structures like mycoplasma cannot be effectively filtered by this technique. It is applicable to sterilize only small quantities.

Commonly used filters in micro biology

The sintered glass filter is made of fused Jen or pyrex glass, manufactured in such a way as to be porous, with pore size and adsorptive charge sufficient to retain bacteria. The seitz filters are compressed asbestos discs having porosity sufficiently small to retain bacteria. Tye chamber land filters are made of porcelain. The mandler/berkfield filters are made of diatomaceous earth. The membrane filter is a cellulose or nitrocellulose membrane with pore size sufficiently small (0.01mm to 10 mm) to trap and thereby remove bacteria from a liquid. The membrane filters are also used to concentrate and trap the micro organisms in water and other liquids. HEPA (High efficiency particulate air filters are of fibre glass filters for sterilization of air.

Low temperature

Most organisms grow very little or not at all at 0° C. Store perishable foods at low temperatures to slow rate of growth and consequent spoilage (eg: milk). Low temperatures are not bactericidal. Psychrotrophs, rather than true psychrophiles are the usual cause of food spoilage in refrigerated foods.

Dessication / Drying (removal of H₂O)

Most micro organisms cannot grow at reduced water activity ($a_w < 0.90$). Often used to preserve foods (eg: fruits, grains etc). methods involve removal of water from product by heat, evaporation, freeze drying, addition of salt or sugar.

Surface tension

is a property of the surface of a liquid that allows it to resist an external force. It is revealed, for example, in floating of some objects on the surface of water, even though they are denser than water, and in the ability of some insects (e.g. water striders) and even reptiles (basilisk) to run on the water surface. This property is caused by cohesion of like molecules, and is responsible for many of the behaviors of liquids.

Surface tension has the dimension of force per unit length, or of energy per unit area. The two are equivalent—but when referring to energy per unit of area, people use the term surface energy—which is a more general term in the sense that it applies also to solids and not just liquids.

In materials science, surface tension is used for either surface stress or surface free energy

Osmotic pressure – plasmolysis/ plasmotysis

Is the process in plant cells where the plasma membrane pulls away from the cell wall due to the loss of water through osmosis. The reverse process, cytolysis, can occur if the cell is in a hypotonic solution resulting in a higher external osmotic pressure and a net flow of water into the cell. Through observation of plasmolysis and deplasmolysis it is possible to determine the tonicity of the cell's environment as well as the rate solute molecules cross the cellular membrane.

Chemical agents

Chemical that is used to kill or inhibit the growth and development of micro organisms are called anti microbial agents. Disinfectants and antiseptics come under anti microbial agents and are usually used on inanimate materials. The mechanism of action is complex and non specific. It may act on lipid portion of cell membrane, oxidize or reduce an important functional group of an enzyme, prevent certain bio synthesis or cause extensive breakdown of DNA.

Types of microbial agents

Chemical sterilants

Chemical sterilants are chemical anti microbial agents that are used for sterilization of heat sensitive substance/ materials. Normally plastic petriplates and medical supplies such as blood transfusion sets, plastic syringes, lenses etc. could be sterilized even in packets or bundles using ethylene oxide, formaldehyde or formalin is effectively used to sterilize enclosed areas/a septic chambers at 22 ° C with a relative humidity of 60 – 80 %.

Antisepetics

Microbicidal agents harmless enough to be applied to the skin and mucous membrane, should not be taken internally. Eg: mercurials, silver nitrate, iodine solution, alcohols, detergents.

Disinfectants

Agents that kill micro organisms, but not necessary their spores, not safe for application to living tissues, they are used on inanimate objects such as tables, floors, utensils etc. eg: Chlorine, hypochlorites, chlorine compounds, Lysol, copper sulfate, quaternary ammonium compounds.

Phenol

Derivative of phenol like benzyl resorcinol, o-cresol, m-cresol, etc., are used as effective disinfectants 5% aqueous solutions of phenols are used as disinfectant. It alters the protein structure and leads to denaturation of proteins and enzymes. Also

affects permeability of cytoplasmic membrane. They readily kill vegetative cells of bacteria and fungi but for spores.

Alcohol

Alcohol at 70% concentration is more effective. It brings about denaturation and coagulation of protein. Ethanol is routinely used in laboratories to surface sterilize worktables and hands of the researcher/ experiment.

Halogens

Halogens such as hypochlorites, choramines and povidone- iodine are used to sanitize utensils, surface sterilize in animate objects, table surfaces and other instruments.

Heavy metals

Heavy metals such as mercuric chloride are also used for surface sterilization purposes. Heavy metals acts as oxidizing agents and kill the micro organisms on the surface of the object. Usually 0.1 % mercuric chloride is used in the laboratories to sterilize the surface of worktable and explants.

Detergents

Detergents are those compounds that make water repellent surfaces more wettable. There are two types of detergents viz., ionic and non ionic. Detergent soaps and other synthetic detergents are used for washing/cleaning glass wares, table tops etc.,

Common antiseptics and disinfectants

Chemical	Action	Uses
Ethanol (50 -70 %)	Denatures proteins and solubilizes lipids.	Anti septic used on skin
Isopropanol (50 – 70 %)	Denatures proteins and solubilizes lipids.	Anti septic used on skin.
Formaldehyde (8%)	Reacts with NH ₂ , SH and COOH groups.	Disinfectant, kills endospores.
Tincture of Iodine (2% in 70 % alcohol)	Inactivates proteins	Antiseptic used on skin
Chlorine (Cl ₂) gas	Forms hypochlorous acid (HClO), a strong oxidizing agent.	Dis infect drinking water, general disinfectant.
Silver Nitrate (Ag NO ₃)	Precipitates proteins.	General antiseptic and

		used in the eyes of newborns.
Mercuric chloride	Inactivates proteins by reacting with sulfide groups.	Disinfectant although occasionally used as an antiseptic on skin.
Detergents (eg: Quaternary ammonium compounds)	Disrupts cell membranes.	Skin antiseptics and disinfectants.

Chemotherapeutic agents

Antimicrobial agents of synthetic origin useful in the treatment of microbial or viral disease. Examples: sulfonilamides, isoniazid, ethambutol, AZT, chloramphenicol.

Antibiotics

Antimicrobial agents produced by micro organisms that kill or inhibit other micro organisms. This is the microbiologist's definition. A more broadened definition of an antibiotic includes many chemical of natural origin which has the effect to kill or inhibit the growth of other types cells. Since most clinically useful antibiotics are produced by micro organisms and are used to kill or inhibit infectious bacteria we follow classic definition.

Antibiotics are low molecular weight (non- protein) molecules produced as secondary metabolites, mainly by micro organisms that live in the soil. Most of these micro organisms form some type of a spore or other dormant cell, and there is thought to be some relationship between antibiotic production and the process of sporulation. Among the molds, the notable antibiotic producers are penicillium and cephalosporium, which are the main source of the beta lactam antibiotics. In the bacteria, the actinomycetes, notable streptomyces species, produce a variety of types of antibiotics including the aminoglycosides (eg: streptomycin), macrolides (eg: erythromycin) and the tetracycline. Endospore forming bacillus species produce polypeptide antibiotics such as polymyxin and bacitracin.

Chemical class	examples	Biological source	Spectrum (effective against)	Mode of action
Beta – lactams (Penicillins and cephalosporins)	Penicillin G, Cephalothin	<i>Penicillium notatum</i> and <i>cephalosporium</i> <i>sp.</i>	Gram positive bacteria	Inhibits steps in cell wall (peptidoglycan) synthesis and

				murein assembly.
Aminoglycosides	streptomycin	<i>Streptomyces griseus</i>	Gram positive and gram negative bacteria	Inhibit translation (protein synthesis)
glycopeptides	vancomycin	<i>Streptomyces orientales</i>	Gram positive bacteria, esp. staphylococcus aureus	Inhibits steps in murein (peptidoglycan) biosynthesis and assembly
macrolides	erythromycin	<i>Streptomyces erythreus</i>	Gram positive and gram negative bacteria not enteric, Neisseria, legionella, mycoplasma	Inhibits translation (protein synthesis)
polypeptides	polymyxin	<i>Bacillus polymyxa</i>	Gram negative bacteria	Damages cytoplasmic membranes
Polyenes	amphotericin	<i>Streptomyces nodosus</i>	Fungi	Inactivate membranes containing sterols
tetracyclines	tetracycline	<i>Streptomyces sp</i>	Gram positive and gram negative bacteria, rickettsias	Inhibit translation (protein synthesis)
Chloramphenicol	Chloramphenicol	<i>Streptomyces venezuelae</i>	Gram positive and gram negative bacteria	Inhibit translation (protein synthesis)